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THE THERMAL CYCLIZATION OF ARENYLPHOSPHONIC ACIDS. II. A NOVEL CYCLODEPHOSPHONATION REACTION

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In the first article in this series² it was shown that the method used by Lynch³ for the preparation of 9,10-dihydro-9-hydroxy-9-phosphaphenanthrene 9-oxide (I -> II) could also be used for the preparation of 4,5-dihydro-4-hydroxy-4-phosphapyrene 4-oxide (III-> IV).

$$\bigcirc \bigcap_{P_0,H_2} \bigcap$$

In an attempt to extend this reaction toward the preparation of the analogous derivative of the 4,9-diphosphapyrene ring system (VI), 2,2'-di(bromomethyl)biphenyl was prepared according to the method of Underwood and Kochmann and converted into the diphosphonate ester by the Arbusov reaction. The ester was hydrolyzed to the diphosphonic acid (V). The diphosphonic acid was then heated in a sublimation apparatus for 24 hours at 330-370°/1.25 mm to effect cyclization. The combined solid material (sublimate and residue) was dissolved in dilute KOH, decolorized, filtered, and acidified with concentrated hydrochloric acid to give a white solid. This was digested two times for 2 hours each with 75 ml of boiling ethanol and filtered. The residue (10% yield), m.p. 392-395° (dec.), was shown to be the expected product (VI).

The diphosphinic acid itself could not be purified sufficiently for analysis; however, treatment of the acid with an excess of diazomethane converted it into an equimolar mixture of the cis and trans dimathyl esters of VI: mp 269-272° (from ethyl acetate); uv max (CHCl₃) 257 mm (c=10,900), 266nm (c=12,800), and 276nm (c=13,300); ir: 1250 cm⁻¹ (P=0) and 1175 and 1030 cm⁻¹ (P=0-CH₃); mmr (CDCl₃): \delta 3.3 (4H, d, J=18, -CH₂-), 3.5 (3H, d, J=11, -OCH₃), 3.7 (3H, d, J=11, OCH₃), 7.1-8.1 (6H, ArH); Anal. Found: C, 57.39; H, 4.82; P, 18.37. Calcd for C₁₆ H₁₆ O₄ P₂: C, 57.49; H, 4.79; P, 18.56. The two nmr doublets at 3.5 and 3.76 correspond to the different methoxyl groups in the cis and trans isomers. Lynch³ has suggested that a uv band at 268nm (c=15,000) in II might arise because the two phenyl rings have been twisted toward a coplanar configuration by formation of the 6-membered ring; the parent acid (I) and its ester show no such maxima between 250-290nm. It should be noted, however, that in the cyclized acid (II) a new Ar-P=0 linkage has been formed which often causes absorption in this area. In either case the assignment of structure VI to our cyclization product is reasonable (the tetraethyl ester of V also shows no such absorption in this area).

The ethanol from the digestion of the cyclization reaction was evaporated to dryness to give a different product (~30%) to which we have assigned structure VII. This was purified by sublimation at 290°/0.6mm to give a product which melted at 246-248°: Anal. Found: C, 68.89; H, 5.24; P, 13.02. Calcd for C₁₄H₁₃O₂P: C, 68.85; H, 5.32; P, 12.71); mmr (CF₃COOD): δ3.2 (4H, d, J=20, -CH₂-), 7.35 (8H, ArH). Diazomethane was also used to convert this compound into its methyl ester: ir: 1240 cm⁻¹ (P=0) and 1190 and 1030 cm⁻¹ (P-0-CH₃). The uv absorption at 250-290nm is absent in this methyl ester (consistent with either the absence of the Ar-P=0 linkage or a less coplanar configuration in the 7-membered ring) and the nmr indicates a ratio of only one -OCH₃ for eight aromatic hydrogens all of which lead to the conclusion that the major product of this cyclization reaction is 6,7-dihydro-6-hydroxy-5H-dibenzo [c,e] phosphepin 6-oxide (VII). The formation of this product suggests that other heterocyclic ring systems might also be prepared by this novel cyclodephosphonation reaction.

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